

Advanced Molecular Modeling Methods

Weds 11am BME 1.112/Zoom

Advances in Modeling Molecular Interactions:

Development of AMOEBA+ Potential – Part I

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AMOEBA: Multipolar and Polarizable Potential

(Atomic Multipole Optimized Energetics for Biomolecular Applications)

$$E_{valence} = E_{bnd} + E_{ang} + E_{strbnd} + E_{opbend} + E_{torsion} + E_{couplings}$$

$$E_{nonbonded} = \sum_{ij} \varepsilon_{ij} \left(\frac{1.07}{\rho_{ij} + 0.07} \right)^7 \left(\frac{1.12}{\rho_{ij}^7 + 0.12} - 2 \right) + \sum (\mathbf{M}_i)^t T_{ij} \mathbf{M}_j + \sum \frac{1}{2} (\mu_i^{ind})^T E_i$$

vdW

Electrostatics
Polarization

$$\mathbf{M}_i = [q_i, \mu_{ix}, \mu_{iy}, \mu_{iz}, Q_{ixx}, Q_{ixy}, \dots, Q_{izz}]^t$$

Ren, *JPCB*, 2003

Ponder, *JPCB*, 2010

Jing, *Ann. Rev. BioPhys.*, 2019

Outlines and Objectives

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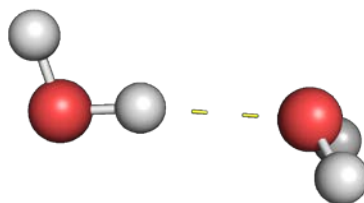
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 - 3.1 Charge Penetration
 - 3.2 Polarization and Charge Transfer
 - 3.3 Geometry Dependent Charge Flux
- 4 Other Improvements
 - 4.1 Polarizability
 - 4.2 van der Waals
 - 4.3 Valence Parameters
- 5 Parametrization

- Theories of SAPT with minimal math
- Practice on running SAPT with Psi4
- How** we explored each energy component
- Lessons learned from the past
- General idea of improving a force field

reading material:

https://biomolmd.org/mw/index.php/Ammm:Mm_aplus

Interaction Energy: *Supermolecular* Approach



$$E_{int} = E_{AB} - E_A - E_B$$

- Gold standard
 - CCSD(T)/CBS
- Basis set superposition error
- ❖ Relies on error cancellation
- ❖ Lack of nature of interactions

Symmetry Adapted Perturbation Theory

- Schrodinger's equations for monomers A and B

$$H_X \Phi_X = E_X \Phi_X, \quad X = A \text{ or } B$$

- Hamiltonian of the dimer

$$H = H_A + H_B + V = H_0 + V.$$

- Solution of the unperturbed H_0

$$E^{(0)} = E_A + E_B \quad \Psi^{(0)} = \Phi_A \Phi_B$$

Symmetry Adapted Perturbation Theory

- The interaction energy expressed as a sum of perturbation corrections

$$E_{\text{int}} = E_{\text{RS}}^{(1)} + E_{\text{RS}}^{(2)} + \dots$$

- Anti-symmetrizer operation

$$\mathcal{A}[a(1)b(2)] = a(1)b(2) - a(2)b(1)$$

- Differences are called exchange energies

$$E_{\text{exch}}^{(i)} = E^{(i)} - E_{\text{RS}}^{(i)}$$

Symmetry Adapted Perturbation Theory

- Hartree-Fock (HF) Approximation and Correlation
 - Mean-field theory
 - Correlation methods (MBPT, CC)

- SAPT Hamiltonian

$$H = F_A + F_B + W_A + W_B + V$$

F : Fock operator

W : Intramonomer correlation operator

V : Intermonomer interaction operator

Symmetry Adapted Perturbation Theory

□ Interaction energy is expressed

$$E_{\text{int}} = \sum_{i=1, j=0} E^{(ij)} \quad i(j) \text{ is the order in } V(W)$$

□ SAPT energy component

$$E_{SAPT} = \sum_{m > 0} \sum_{n \geq 0} E_{SAPT}^{(mn)} = E_{elst} + E_{exch} + E_{ind} + E_{disp}$$

Symmetry Adapted Perturbation Theory

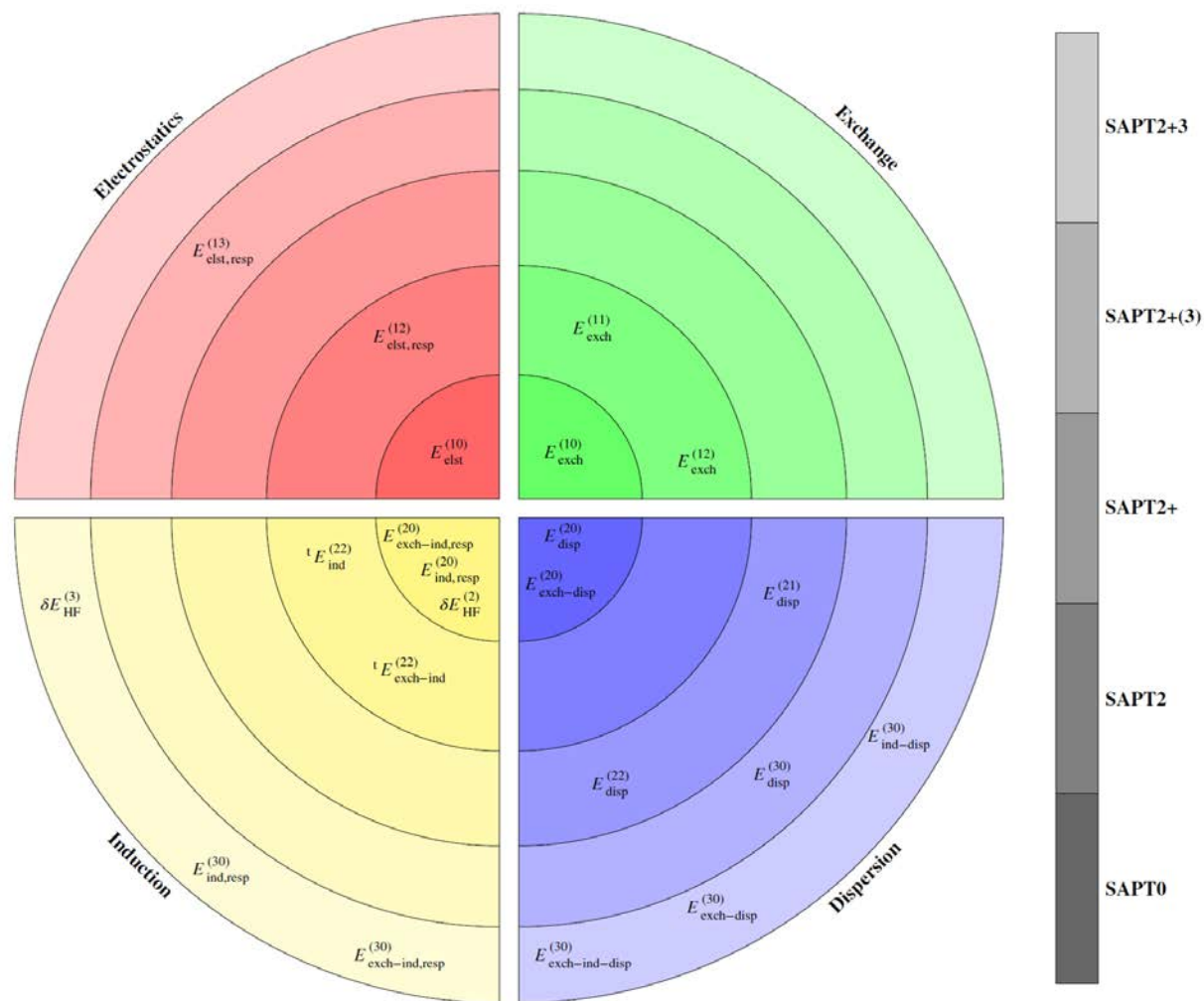
□ Levels of SAPT

$$\begin{aligned}
 E_{SAPT0} &= E_{elst}^{(10)} + E_{exch}^{(10)} + E_{ind,resp}^{(20)} + E_{exch-ind,resp}^{(20)} + \delta E_{HF}^{(2)} + E_{disp}^{(20)} + E_{exch-disp}^{(20)} \\
 E_{SAPT2} &+= E_{elst,resp}^{(12)} + E_{exch}^{(11)} + E_{exch}^{(12)} + {}^tE_{ind}^{(22)} + {}^tE_{exch-ind}^{(22)} \\
 E_{SAPT2+} &+= E_{disp}^{(21)} + E_{disp}^{(22)} \\
 E_{SAPT2+(3)} &+= E_{elst,resp}^{(13)} + E_{disp}^{(30)} \\
 E_{SAPT2+3} &+= E_{exch-disp}^{(30)} + E_{ind-disp}^{(30)} + E_{exch-ind-disp}^{(30)}
 \end{aligned}$$

Symmetry Adapted Perturbation Theory

□ Levels of SAPT

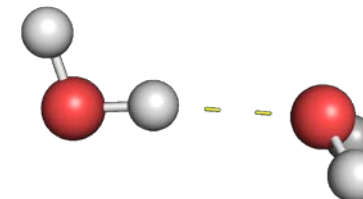
- Electrostatics
- Induction
- Exchange
- Dispersion



SAPT and CCSD(T)/CBS Calculations

□ Input files

- `/home/liuchw/public_html/AMM_aplus`
- https://biomol.bme.utexas.edu/~liuchw/AMM_aplus



□ Compare the results

SAPT and CCSD(T)/CBS Calculations

SAPT

SAPT Results			
Electrostatics	-12.88305289 [mEh]	-8.08423774 [kcal/mol]	-33.82445069 [kJ/mol]
Elst10,r	-13.28732680 [mEh]	-8.33792345 [kcal/mol]	-34.88587172 [kJ/mol]
Elst12,r	0.16858266 [mEh]	0.10578721 [kcal/mol]	0.44261370 [kJ/mol]
Elst13,r	0.23569126 [mEh]	0.14789850 [kcal/mol]	0.61880732 [kJ/mol]
Exchange	12.88899409 [mEh]	8.08796590 [kcal/mol]	33.84004933 [kJ/mol]
Exch10	11.14311571 [mEh]	6.99241068 [kcal/mol]	29.25624627 [kJ/mol]
Exch10(S^2)	11.06411493 [mEh]	6.94283694 [kcal/mol]	29.04882975 [kJ/mol]
Exch11(S^2)	0.14286433 [mEh]	0.08964872 [kcal/mol]	0.37509025 [kJ/mol]
Exch12(S^2)	1.60301405 [mEh]	1.00590650 [kcal/mol]	4.20871280 [kJ/mol]
Induction	-4.02156434 [mEh]	-2.52356972 [kcal/mol]	-10.55861572 [kJ/mol]
Ind20,r	-4.77703438 [mEh]	-2.99763433 [kcal/mol]	-12.54210203 [kJ/mol]
Ind22	-0.70315409 [mEh]	-0.44123585 [kcal/mol]	-1.84613081 [kJ/mol]
Exch-Ind20,r	2.61902451 [mEh]	1.64346270 [kcal/mol]	6.87624792 [kJ/mol]
Exch-Ind22	0.38550650 [mEh]	0.24190898 [kcal/mol]	1.01214718 [kJ/mol]
delta HF,r (2)	-1.46694123 [mEh]	-0.92051952 [kcal/mol]	-3.85145368 [kJ/mol]
delta MP2,r (2)	-0.07896565 [mEh]	-0.04955169 [kcal/mol]	-0.20732429 [kJ/mol]
Dispersion	-3.85052017 [mEh]	-2.41623789 [kcal/mol]	-10.10953932 [kJ/mol]
Disp20	-4.05563051 [mEh]	-2.54494657 [kcal/mol]	-10.64805645 [kJ/mol]
Disp30	0.14638417 [mEh]	0.09185745 [kcal/mol]	0.38433159 [kJ/mol]
Disp21	0.07191071 [mEh]	0.04512465 [kcal/mol]	0.18880155 [kJ/mol]
Disp22 (SDQ)	-0.06804450 [mEh]	-0.04269857 [kcal/mol]	-0.17865081 [kJ/mol]
Disp22 (T)	-0.57474478 [mEh]	-0.36065779 [kcal/mol]	-1.50899220 [kJ/mol]
Est. Disp22 (T)	-0.68081841 [mEh]	-0.42722000 [kcal/mol]	-1.78748849 [kJ/mol]
Exch-Disp20	0.73567837 [mEh]	0.46164515 [kcal/mol]	1.93152329 [kJ/mol]
Total HF	-5.76916219 [mEh]	-3.62020393 [kcal/mol]	-15.14693324 [kJ/mol]
Total SAPT0	-9.08911433 [mEh]	-5.70350535 [kcal/mol]	-23.86346639 [kJ/mol]
Total SAPT2	-7.49230089 [mEh]	-4.70148979 [kcal/mol]	-19.67103327 [kJ/mol]
Total SAPT2+	-8.16925309 [mEh]	-5.12628371 [kcal/mol]	-21.44837102 [kJ/mol]
Total SAPT2+(3)	-7.78717766 [mEh]	-4.88652775 [kcal/mol]	-20.44523212 [kJ/mol]
Total SAPT2+dMP2	-8.24821874 [mEh]	-5.17583540 [kcal/mol]	-21.65569531 [kJ/mol]
Total SAPT2+(3)dMP2	-7.86614331 [mEh]	-4.93607945 [kcal/mol]	-20.65255641 [kJ/mol]
Special recipe for scaled SAPT0 (see Manual):			
Electrostatics sSAPT0	-13.28732680 [mEh]	-8.33792345 [kcal/mol]	-34.88587172 [kJ/mol]
Exchange sSAPT0	11.14311571 [mEh]	6.99241068 [kcal/mol]	29.25624627 [kJ/mol]
Induction sSAPT0	-3.56844793 [mEh]	-2.23923488 [kcal/mol]	-9.36895874 [kJ/mol]
Dispersion sSAPT0	-3.30408052 [mEh]	-2.07334183 [kcal/mol]	-8.67486222 [kJ/mol]
Total sSAPT0	-9.01673954 [mEh]	-5.65808949 [kcal/mol]	-23.67344641 [kJ/mol]

CCSD(T)

==> N-Body: Counterpoise Corrected (CP) energies <==

n-Body	Total Energy [Eh]	I.E. [kcal/mol]	Delta [kcal/mol]
1	N/A	0.000000000000	0.000000000000
2	N/A	-5.112334965607	-5.112334965607

$$E_{HF} = -3.62 \text{ kcal/mol}$$

$$E_{SAPT0} = -5.70 \text{ kcal/mol}$$

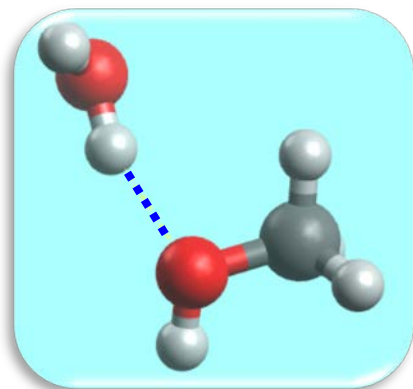
$$E_{SAPT2+} = -5.13 \text{ kcal/mol}$$

$$E_{SAPT2+dMP2} = -5.18 \text{ kcal/mol}$$

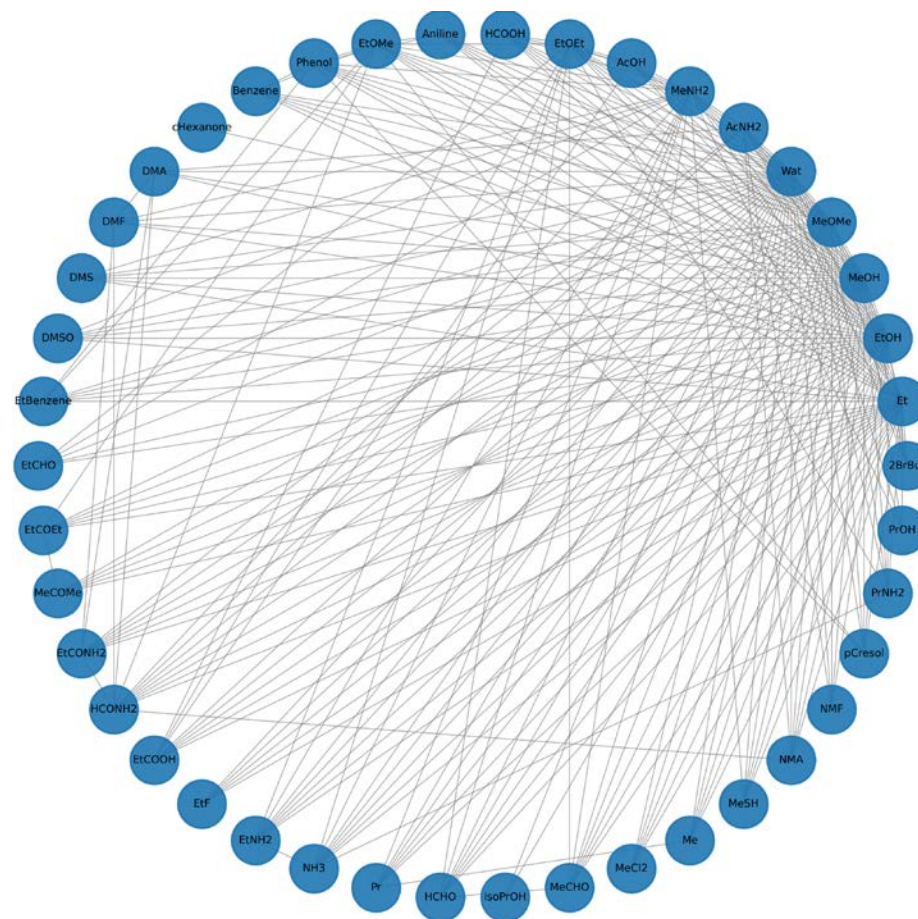
$$E_{SAPT2+(3)dMP2} = -4.94 \text{ kcal/mol}$$

$$E_{CCSD(T)/CBS} = -5.11 \text{ kcal/mol}$$

Dimer Interaction Networks for Organics



(Wat-MeOH)



- **Dimer set**
 - Homodimer
 - Heterodimer
 - 42 molecules
 - 280 pairs
 - 10 separations
- **Interaction energy**
 - SAPT2+(3)dMP2/aTZ
 - MP2/CBS
 - CCSD(T)/CBS (some)

- Element coverage: C, H, O, N, S, F, Cl, Br